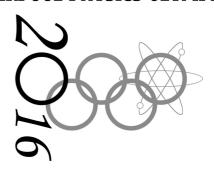
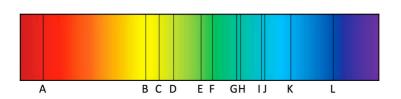
LIVERPOOL PHYSICS OLYMPICS

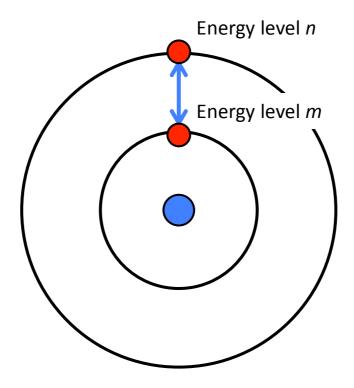




Stellaments

Aim	To identify five absorption lines in the spectrum of a star.			
Materials	1 stellar spectrum	1 m rule		
Method	Use the position of line 'F' in the spectrum, originating from transitions in hydrogen atoms between energy levels 2 and 4 (see over) to find the letters that identify the transitions from:			
	He (4 → 7) Li (4 → 5)	Be (6→8)	B (6 → 7)	C (7 → 8)
Conditions	The five letters must be given in the correct order.			
Time Limit	25 minutes			
Ranking	The ranking order will be determined by the time taken to correctly identify the five lines in the spectrum.			
Team			Result	
Do not write belo	ow this line			
			Rank	,

Bohr model of the atom



An electron transition between energy levels n and m of an atom with atomic number Z will absorb or emit a photon with energy E given by

$$E \propto Z^2 \left(\frac{1}{n^2} - \frac{1}{m^2} \right)$$